$R(\alpha)=R_{0}\left\{1+\mathrm{r} \frac{\left[3 \cos ^{2} \alpha-(1+F+G-2 F G)\right]}{(1+F)(1+G)}\right\}$ where $R(\alpha)$ is the recorded signal, and $R_{0}$ is the signal that would be recorded for an equivalent but depolarized sample (i.e. $r=0$ ) and thus is directly proportional to the total intensity emitted by the sample.

For a given $\alpha ; r, F, G$, and $R_{0}$ can be obtained by placing polarizers in the excitation and emission beams and taking four orthogonal readings $R_{V}^{V}(\alpha), R_{\mathrm{H}}^{\mathrm{V}}(\alpha), R_{V}^{\mathrm{H}}(\alpha)$ and $R_{H}(\alpha)$ where superscripts refer to the mode of the excitation polarizer and subscripts refer to the orientation of the emission polarizer. V refers to vertical and H to horizontal where the plane of the instrument is in the horizontal (see Ref. 1 for procedure).

If a measure of $\boldsymbol{R}_{\mathbf{0}}$ only is desired, it has been found that if a polarizer is placed in the emission beam at an angle of $5434^{\circ}$ $\left(\cos ^{-1} \sqrt{ } 1 / 3\right)$ from the vertical, and if the emission is viewed at $45^{\circ}$ to the direction of propagation of the exciting light, the reading obtained is
$R_{54.75}\left(45^{\circ}\right)=\frac{1}{3} \quad R_{\mathrm{O}}\left(\frac{G+2}{G+1}\right)$
Thus this reading is independent not only of the state of polarization of the exciling light (i.e. $F$ ) but also the degree of emission anistropy $r$, and is also directly proportional to $R_{0}$.

These equations have been verified experimentally with data obtained from a sample which has a high degree of emission anistropy (Nile Blue A perchlorate in glycerol). A recently designed goniospectrofluorimeter was used and will be described.

1 E. D. Cehelnik, K. D. Mielenz and R. A. Velapoldi, J. Res. N. B. S., 79A (1975) 1.
2 K. D. Mielenz, E. D. Cehelnik and R. L. McKenzie, to be submitted, J. Chem. Phys., (1975).

## G4 <br> Excited State Relaxation: Laser Pumped Fluorescence Studies

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We describe recent measurements of fluorescence with simultaneous wavelength
and time resolution of emission. The laser source is a mode-locked, cavity dumped dye laser pumped by an $\mathrm{Ar}^{+}$laser. The tunable visible light is converted to ultraviolet with non-linear optical crystals and the output pulse train consists of subnanosecond pulses separated by 200 ns or longer intervals. The method of time-correlated photon counting allows high sensitivity in studies of single vibronic (and even rovibronic) excitation of fluorescence, fluorescence quenching, and excited state vibrational relaxation. Results will be presented for a variety of molecules including ketones and aza-aromatics.

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## G7

## Energy Transfer From Single Vibronic Levels of Benzene ( ${ }^{\mathbf{1}} \mathbf{B}_{\mathbf{2 u}}$ ) and Fluorobenzene ( ${ }^{1} \mathrm{~B}_{1}$ )

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Electronic and vibrational energy transfer from specific vibronic levels of the first excited singlet states of benzene and fluorobenzene to a variety of acceptor molecules has been investigated. The excited states, prepared with a narrow band excitation source at low pressures ( $<0.3$ Torr), are monitored by both individual fluorescence transitions and total fluorescence from the donor molecule. The energy tranfer from these states can be separated into effective collison cross sections for electronic ( $\sigma_{e}^{2}$ ) and vibrational ( $\sigma_{v}^{2}$ ) energy transfer.

Examination of quenching by molecules capable of both electronic and vibrational energy removal from vibronic states of benzene and those capable of only vibrational energy removal indicate that $\sigma_{v}^{2}$ is comparable for the series of molecules acetone, acetone- $d_{6}$, isopropyl alcohol, methylether and propane. Further $\sigma_{v}^{2}$ is found to be linearly dependent on excess vibrational energy ( $\epsilon_{\mathrm{v}}$ ) in the excited electronic state. As $\epsilon_{\mathrm{v}}$ varies from 0 to $2300 \mathrm{~cm}^{-1}, \sigma_{\mathrm{v}}^{2}$ varies from $4 \AA^{2}$ to $47 \AA^{2}$. Previous estimates [1] of $\sigma_{\mathrm{e}}^{2}$ have been verified and indicate that

